

Is sodium sulfate decahydrate a phase change energy storage material?

In this paper, sodium sulfate decahydrate (SSD) with a phase transition temperature of 32 °C was selected as the phase change energy storage material. However, SSD has the problems of large degree of supercooling, obvious phase stratification, and low thermal conductivity.

Is sodium sulfate a good salt hydrate?

For example, sodium sulfate decahydrate, $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$ (SSD), has been identified as one of the most promising salt hydrates for building applications due to its low cost (1.60 \$/kWh), high energy storage capacity (254 J/g), and moderate melting temperature (32.4 °C) [20,21].

Why is sodium sulfate decahydrate problematic?

Sodium sulfate decahydrate has been studied due to its high supercooling, phase separation, and low thermal conductivity, which greatly limit the practical application of sodium sulfate decahydrate [45,46].

What is the melting temperature of sodium sulfate?

The melting temperature of sodium sulfate in the composite materials is around 880 °C and no confinement effect is observed due to the nanoscale diatomite pore size. This indicates that the material should be used at temperatures over 890 °C to maximize the energy density.

Does sodium sulfate decahydrate affect thermal conductivity?

In the samples SSD-BCKN3, SSD-BCKN3-1, SSD-BCKN3-2, SSD-BCKN3-3, and SSD-BCKN3-4, with the same proportion of added materials, the thermal conductivity increased with the decrease of sodium sulfate decahydrate content. The standard deviations were mostly similar.

What happens when sodium sulfate (SS) melts?

When SSD undergoes melting, the water molecules from the crystalline SSD phase are released to the liquid water phase. The remaining anhydrous sodium sulfate (SS) salt cannot be fully dissolved in the water, resulting in a saturated salt solution with undissolved salt particles (Fig. 1 b and f).

Study on performance optimization of sodium sulfate decahydrate phase change energy storage materials Xian Dong 1 ¹; Jinfeng Mao 1 ¹; Shibin Geng 1 ¹; Yong Li 1 ¹; Pumin Hou 1 ¹; Huiliang Lian 1

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These results demonstrate the potential of as-prepared microencapsulated SSD composite phase-change

energy storage materials for cooling water applications. Discover the ...

It appears that the major problem preventing use of sodium sulfate decahydrate for thermal energy storage can be avoided by using the composition which is at or slightly to the water-rich side of the invariant point in the phase diagram. A mixture of 68.2 w/o Na₂SO₄ · 10H₂O and 31.8 w/o H₂O is suggested for a TES material.

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In this paper, the system Li₂SO₄-Na₂SO₄ is proposed as a candidate material for thermal energy storage applications at high temperatures (450-550 °C). Depending on the composition, the thermal energy can be stored by using a eutectoid reaction and solid-solid phase transition.

Cooling experiments and fruit storage performance experiments showed that SSD-BCKN3 has good potential for energy storage in cold chain transportation applications.

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Telkesd et al. [1,2,3] extensively studied the PCMs for energy storage applications and, for the first time, established an energy storage solar house using sodium sulfate decahydrate (Na₂SO₄ · 10H₂O) as a PCM. Since the 1970s, the theoretical understanding and utilization of heat storage technology have gained significant research ...

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J/g)????????????????,????????????????,????????????????? ...

This work explores the use of sodium sulfate and diatomite to formulate composite materials for high temperature thermal energy storage applications. Sodium sulfate in the composite functions as a phase change material (PCM) and diatomite as a structural skeleton for shape stabilization.

Most energy analysts believe LIB will capture most all energy-storage growth in stationary/transportation markets over the next 10 years. The appeal of LIB storage is its continuous price drop, flexible installation (including modular stacking), fast response, and short construction time.

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